A Convenient Synthesis of Enamides and Dienamides by Horner-Wittig and Wadsworth-Emmons Reactions

Axel Couture,* Eric Deniau and Pierre Grandclaudon

Laboratoire de Chimie Organique Physique associé au CNRS, Université des Sciences et Technologies de Lille, 59650 Villeneuve d'Ascq, France

Abstract: Various N-acyl-N-alkyl-1-amino-alkenes and -1,3-dienes have been efficiently prepared by reacting aldehydes or ketones 3 with N-alkyl-N-(diphenylphosphinoyl)methyl and -N-(diethoxyphosphoryl)methyl carboxamides 1, 2.

N-acyl-N-alkyl-1-amino-alkenes and -1,3-dienes are a class of conjugated compounds of increasing interest in organic synthesis.¹ For instance, the photocyclization of aromatic enamides has been established as one of the most useful cyclization reactions for constructing six-membered lactams² and has been remarkably exploited in the area of alkaloid total synthesis.³ In the last few years, the use of dienamides⁴ in both inter- and intramolecular Diels-Alder reactions has been impressively exemplified,⁵ especially for the elaboration of natural products.⁶ Recent reports from our laboratory showed the utility of (poly)cyclic dienamides for the construction of diversified spirooxazines, thiazines and selenazines.⁷

The main methods for the preparation of these conjugated compounds involve acylation of aldimines⁸ and vinylimines⁹ with carboxylic acid chlorides in the presence of a tertiary amine or the multistep reaction sequence: anionisation, alkylation and basic elimination from α -carbamidosulfones.¹⁰ A convenient preparation of 1-N-acylamino-1,3-dienes from dienoic acids by a modified Curtius procedure has been reported but this route leads exclusively to insaturated carbamates.¹¹ Very recently, Palomo and coworkers described a new method for the preparation of diversely substituted enamides through fluoride ion mediated Peterson alkenation of N-[C,C-bis(trimethylsilyl)methyl]amido derivatives.¹²

In this paper we report that a variety of N-acyl-N-alkyl-1-amino-alkenes and -1,3-dienes 5-12 can be efficiently prepared by Horner-Wittig and Wadsworth-Emmons reactions of N-(diphenylphosphinoyl)methyl or N-(diethoxyphosphoryl)methyl carboxamides, 1, 2 respectively, with suitable aldehydes and ketones 3 (scheme 1, Table). A rather similar strategy has been successfully utilized to convert aromatic and aliphatic aldehydes into their homologous enamines.¹³

The N-alkyl-N-(diphenylphosphinoyl)methyl and -N-(diethoxyphosphoryl)methyl carboxamides, 1 and 2 were prepared by treatment of the crude chloromethylcarboxamides 14 ¹⁴ with ethyl diphenylphosphinite 15 ¹⁵ or triethylphosphite 16 (scheme 1). ¹⁶ This method proved more convenient than the treatment by chlorodiphenylphos-

scheme 1

phine of N-(α -alkoxyalkyl)-amides which can only be prepared by a few limited methods ¹⁷ such as anodic oxidation of N-alkylamides. ¹⁸ Various N-(1-diphenylphosphinoyl)alkyl carboxamides have been also synthesized by α -amidoalkylation of chlorodiphenylphosphine but the procedure was limited. ¹⁹ N-(diethoxyphosphoryl)methyl carboxamides can only be prepared by dehydrogenation with Hg(II)-EDTA of dialkylaminomethylphosphonates ²⁰or by acylation of arylaminomethylphosphonates. ²¹

	R ¹	R ²	R ³	yield% (m.p.°C)		R ¹	R ²	R ³	yield %	_
1a	Ph	Me	Ph	ref. 15	2a	Ph	Ме	OEt	90	
1b	2-furyl	Me	Ph	90 (118-119)	2b	2-furyl	Ме	OEt	87	
1c	benzyl	Ме	Ph	ref. 15	2d	-(C	H ₂) ₃ -	OEt	88	
1d	-(CH ₂)	3	Ph	ref. 15			Ū			
1e	Ph	benzyl	Ph	ref. 15						

scheme 2

The phosphorylated amides 1 and 2 were smoothly deprotonated at $-78\,^{\circ}$ C with lithium diisopropylamide in THF to give an orange solution of the corresponding anions which were subsequently treated with a variety of aromatic, aliphatic aldehydes and ketones and with α , β -unsaturated aldehydes 3 (scheme 1). Refluxing the reaction mixture for a short period (0.5 h) was necessary to insure completion of the reaction. Results are compiled in Table where it may be seen that this simple procedure affords N-acyl-N-alkyl-1-amino-alkenes and -1,3-dienes, 5-12 in good yields.

It is noteworthy that the resulting unsaturated carboxamides obtained from the lithiated Horner-Wittig reagents 1 and aromatic and α , β -unsaturated aldehydes were exclusively obtained as E-or E, E-isomers. However a much lower degree of stereoselectivity was observed with aliphatic aldehydes and with the phosphonic acid esters 2. All these results are consistent with previous reported findings. The stereoselectivity was also sensitive to the metal counterion in the adducts 4. Thus the treatment of the phosphorylated carboxamides 1a and 2a with potassium bis(trimethylsilyl)amide (KHMDS)²⁴ at -30 °C in THF followed by the addition of p-anisaldehyde led, after warming to room temperature and usual work-up, to an approximate 75/25 mixture of Z-/E-isomers (as judged by NMR).

Table.	Enamides	5-10	and	Dienamides	11	,12	Prepared	

Entry	R ¹	R ²	R ⁴	R ⁵	m.p. (°C)*	yield% (<i>E/Z</i>)		
					E-isomer	from 1	from 2	
5a	Ph	Me	Ph	Н	101-102	92(100/-)	88(90/10)	
5b	2-furyl	Me	Ph	H	56-57	90(100/-)	87(90/10)	
5c	benzyl	Me	Ph	H	79-80	78(100/-)		
5d	-(CH ₂)	-	Ph	H	117-118	85(100/-)	80(95/5)	
ба	Ph	Me	p-MeO-C ₈ H ₄	H	94-95	91(100/-)	87(90/10)	
6b	2-furyl	Me	p-MeO-C ₅ H ₄	н	52-53	90(100/-)	85(90/10)	
7a	Ph	Me	Me	H		45(50/50)	41(60/40)	
8a	Ph	Me	cyclohexyl	H		65(60/40)	55(65/35)	
8b	2-furyl	Me	cyclohexyl	H		51(60/40)		
9a	Ph	Me	benzyl	Н		40(65/35)	41(65/35)	
10a	Ph	Me	Ph	Ph	82-83	92		
10e	Ph	benzyl	-((CH ₂).	85-86 ²⁷	48		
11a	Ph	Me	CH=CH-Ph	Н	126-127	88(<i>E</i> , <i>E</i> -isom	er)	
11b	2-furyl	Me	CH=CH-Ph	Н	92-93	85(<i>E,E</i> -isom	er)	
12a	Ph	Me	$CH = C(Me)_2$	Н		83(E,E-isomer)		
12b	2-furyl	Me	$CH = C(Me)_2$	Н		80(<i>E</i> , <i>E</i> -isom	er)	

^{*} All new compounds gave satisfactory ¹H NMR, mass and IR spectra.

In summary, the procedure described here provides a convenient, simple and versatile method for the preparation of enamides and dienamides from easily accessible Horner-Wittig and Wadsworth-Emmons reagents.

References and Notes

- 1. For a review, see: Campbell, A.L.; Lenz, G.R. Synthesis 1987, 421.
- 2. a) Naito, T.; Tada, Y.; Nishiguchi, Y.; Ninomiya, I. J. Chem. Soc., Perkin Trans. I 1985, 487 and cited references. b) Couture, A.; Grandclaudon, P.; Hooijer, S. J. Org. Chem. 1991, 56, 4977.
- Ninomiya, I.; Naito, T. "The Alkaloids"; Brossi, A. Ed.; Academic Press: New York, 1983; Vol. XXII, pp. 189-279.
- 4. For reviews, see: a) Smith, M.B. Org. Prep. Proc. Int. 1990, 22, 315. b) Weinreb, S.N., "Hetero Diels-Alder Methodology in Organic Synthesis"; Wasserman, H.H. Ed.; Academic Press: New York, 1987.
- 5. Petrzilka, M.; Grayson, J.I. Synthesis 1981, 768.
- a) Oppolzer, W.; Fröstl, W. Helv. Chim. Acta 1975, 58, 587 and 590. b) Oppolzer, W.; Fröstl, W.;
 Weber, H.P. Helv. Chim. Acta 1975, 58, 593. c) Oppolzer, W.; Flaskamp, E. Helv. Chim. Acta 1977, 60,
 204. d) Oppolzer, W. Angew. Chem., Int. Ed. Engl. 1977, 16, 10.
- 7. Couture, A.; Bochu, C.; Grandclaudon, P. J. Org. Chem. 1988, 53, 4852 and cited references.
- 8. Breederveld, M. Recl. Trav. Chim. Pays-Bas 1960, 79, 1197.
- 9. Oppolzer, W.; Bieber, L.; Francotte, E. Tetrahedron Lett. 1979, 4537.
- 10. Berthon, L.; Uguen, D. Tetrahedron Lett. 1985, 3975.

- 11. Overman, L.E.; Taylor, G.F.; Petty, C.B.; Jessup, P.J. J. Org. Chem. 1978, 43, 2164.
- 12. Palomo, C.; Aizpura, J.M.; Legido, M.; Picard, J.P.; Dunogues, J.; Constantieux, T. *Tetrahedron Lett.* 1992, 3903.
- a) Böhme, H.; Haake, M.; Austerhoff, G. Arch. Pharm. Chem. 1972, 305, 88. b) Martin, S.F. J. Org. Chem. 1974, 39, 2814. c) Martin, S.F. Synthesis 1979, 633. d) Martin, S.F. Tetrahedron 1980, 36, 419 and cited references. e) Broekhof, N.L.J.M.; Jonkers, F.L.: van der Gen, A. Tetrahedron Lett. 1979, 2433. e) Broekhof, N.L.J.M.; Jonkers, F.L.; van der Gen, A. Tetrahedron Lett. 1980, 2671. g) Broekhof, N.L.J.M.; van Elburg, P.; van der Gen, A. Recl. Trav. Chim. Pays-Bas 1984, 103, 312; Broekhof, N.L.J.M.; van der Gen, A. ibid. 305; Broekhof, N.L.J.M.; van Elburg, P.; Hoff, D.J.; van der Gen, A. ibid. 317.
- 14. Couture, A.; Deniau, E.; Grandclaudon, P. Synthetic Commun. 1992, 22, 2381.
- 15. Rabinowitz, R.: Pellon, J. J. Org. Chem. 1961, 26, 4623.
- 16. Typical procedure for the synthesis of 1 and 2. A mixture of the amides (30 mmol), paraformaldehyde (900 mg) and Me₃SiCl (9.8 g, 90 mmol) in CHCl₃ (150 mL) was refluxed for 6 h. The mixture was filtered. The solvent and the slight excess of paraformaldehyde were removed under vacuo (5x10⁻²Torr). The crude chloromethyl carboxamides were dissolved in toluene and were subsequently treated with ethyl diphenylphosphinite 15 (for 1a-e) or triethyl phosphite 16 (for 2a,b,d). The mixture was refluxed for 0.5 h and the solvent removed in vacuo. Compounds 1a-e were obtained by trituration of crude products with ether, filtration and recrystallization from hexane/toluene. Compounds 2a,b,d were purified by flash column chromatography on silica using a mixture acetone/hexane (7:3) as eluent.
- For preprations of N-(α-alkoxyalkyl)amides see: Katritzky, A.R.; Fan, W-Q.; Black, M.; Pernak, J. J. Org. Chem. 1992, 57, 547.
- 18. Shono, T. "Electroorganic Synthesis"; Academic Press: London, 1991; pp. 71-74.
- 19. Oleksyszyn, J. Synthesis 1981, 444.
- 20. Mohrle, H.; Vetter, W. Z. Naturforsch. 1988, 43b, 1662.
- 21. Lukanov, L.K.; Venkov, A.P. Synthesis 1992, 263.
- 22. Selected H NMR spectra (CDC1, 400 MHz): 6a(E) δ 3.35 (s, NCH₃), 3.77 (s, OCH₃), 5.99 (d, J= 14.4 Hz, ArCH=), 7.08 (b.s, NCH=), 6.70-7.90 (m, 9Haryl). 6b(E) δ 3.41 (s, NCH₃), 3.81 (s, OCH₃), 6.09 (d, J= 14.5 Hz, ArCH=), 6.54 (dd, J= 0.8, 3.5 Hz, Hfuran), 7.10 (dd, J= 0.8, 3.5 Hz, Hfuran), 7.60 (dd, J= 0.8, 1.8 Hz, Hfuran), 7.89 (d, J= 14.5 Hz, NCH=), 6.70-8.10 (m, 4Haryl). 11a(E,E) δ 3.31 (s, NCH₃), 5.91 (dd, J= 10.4, 13.4 Hz, ArCH=CH), 6.49 (d, J= 13.4 Hz, ArCH=), 6.78 (bs, NCH=CH), 7.10-7.65 (m, 9Haryl+NCH=). 11b(E,E) δ 3.39 (s, NCH₃), 5.99 (dd, J= 10.5, 13.8 Hz, ArCH=CH), 6.51 (dd, J= 1.8, 3.5 Hz, Hfuran), 6.53 (d, J= 13.8 Hz, ArCH=), 6.80 (dd, J= 10.5, 15.5 Hz, NCH=CH), 7.10 (dd, J= 0.8, 3.5 Hz, Hfuran), 7.38 (dd, J= 0.8, 1.8 Hz, Hfuran), 7.59 (d, J= 15.5 Hz, NCH=), 7.10-7.70 (m, 5Haryl).
- 23. a) Edmundson, R.S. "The Chemistry of Organophosphorus Compounds"; Hartley, F.R., Ed.; John Wiley and Sons: Chichester, 1992; Vol. 2, pp; 287-409. b) Buss; A.D.; Warren, S. J. Chem. Soc., Perkin Trans. I 1985, 2307. c) Maryanoff, B.E.; Reitz, A.B. Chem. Rev. 1989, 89, 863.
- 24. Brown, C.A. Synthesis 1974, 427.
- 25. The E- and Z-isomers were separated by flash column chromatography using a mixture AcOEt/hexane (1:1) as eluent. H NMR spectrum of 6a(Z) δ 3.08 (s, NCH₃), 3.78 (s, OCH₃), 5.78 (d, J= 8.6 Hz, ArCH=), 6.17 (d, J= 8.6 Hz, NCH=), 6.85-7.55 (m, 9Haryl).
- 26. The E-isomers could not be obtained by thermal conversion of the Z-isomers in contrast to results reported for enamines synthesis (ref. 13g p. 305).
- 27. Gramain, J.C.; Mavel, S.; Troin, Y.; Vallee-Goyet, D. Tetrahedron 1991, 47, 7287.